

# Public Viewing Syllabus for Earth Sciences 4431a (v. 1.0, July 10, 2025)

# **1.** Course Information

### Earth Sciences 4431a. Stable Isotope Geochemistry in Earth & Environmental Sciences

- Fall Term
- In person

#### **List of Prerequisites**

- Earth Sciences 2230A/B or 3341A/B or completion of any 2000 level half-course in Chemistry; or registration in the third or fourth year of an Environmental Sciences module; or permission of the Instructor/ Department/Faculty.
- Unless you have either the prerequisites for this course or written special permission from the Department of Earth Sciences or their designate to enroll in it, you may be removed and withdrawn from this course in accordance with university policy. This may be done after the add/drop deadline of the academic term, and the course will be marked as withdrawn (WDN) on your academic record. This decision may not be appealed.

# 2. Instructor Information

- Instructor: Fred J Longstaffe, Dept. Earth Sciences
- Graduate Teaching Assistant: To be announced, to be announced
- Students must use their Western (@uwo.ca) email addresses when contacting their instructors.
- Weekly office hours with Graduate Teaching Assistant:
  - *To be announced*; format virtual

To be announced; format in person

When known, details will be provided via the OWL Brightspace course site.

• office hours with Instructor – always welcome by appointment

# 3. Course Syllabus, Schedule, Delivery Mode

*Lay Summary:* You will learn that you are what you eat – isotopically. You will learn that your hair and teeth retain isotopic signals of where you have lived. You will learn that it is not all about you. Minerals have a history too, and you will learn how to decode that history using stable isotopes, in terms of temperatures and fluids involved in mineral formation. You will learn what sorts of water-rock-organic interaction might make you rich and at what cost to Earth. Stable isotopes hold the key to

understanding much of past climate change, paleo-ecological shifts and extinctions. You will learn to think like an *Isotopist*. Your life will never be the same thereafter.

*Technical Summary:* Atoms of many elements come in different varieties known as isotopes. Isotopes of an atom have the same number of protons and electrons, but a different number of neutrons. Stable isotopes of an atom do not decay to form another element. Because of the difference in mass arising from different numbers of neutrons, stable isotopes of a given element behave ("fractionate") in slightly different ways during reactions such as (i) evaporation of water, (ii) carbon dioxide fixation during photosynthesis, (iii) crystallization of a magma, and (iv) transfer of a donut's isotopic signature to your fingernails and hair. This course addresses the principles governing the fractionation of stable isotopes, and focuses on how the stable isotopes of oxygen, hydrogen, carbon, nitrogen and sulphur allow us to trace interactions within the atmosphere-hydrosphere-biosphere-lithosphere Earth System.

### Learning Outcomes

Upon successful completion of this course students will be able to:

- 1. Use the oxygen- and hydrogen-isotope compositions of water to determine its source and the hydrological processes that have shaped the water's isotopic composition, as evaluated through assignments, tests, and oral and written examinations
- 2. Identify the sources (mantle, crust) that have contributed to magma generation, and recognize the nature of rock-water interaction that may have affected an igneous rock after its crystallization, as evaluated through assignments, tests, and oral and written examinations
- 3. Calculate the stable isotope fractionation factor between two phases (e.g., mineral and water), use these data to establish whether the system is in equilibrium, and for equilibrium systems, determine the temperature at which equilibration occurred, as evaluated through assignments, a written test, and oral and written examinations
- 4. Use the oxygen- and hydrogen-isotope composition of clay minerals to determine conditions of weathering and hydrothermal alteration, and interpret these results within the larger framework of diagenesis, ore mineralization and / or climate change over Earth history, as evaluated through assignments, a written test, and oral and written examinations
- 5. Identify photosynthetic pathways in vegetation using stable carbon-isotope compositions, recognize land use changes based on stable carbon-isotope compositions of organic matter, and determine the diet and trophic level of animals based on the stable carbon- and nitrogen-isotope compositions of their tissues, as evaluated through a written test, and oral and written examinations
- 6. Use the stable carbon- and oxygen-isotope compositions of Earth's biosphere, atmosphere and hydrosphere to identify changes in Earth's carbon and water cycles at geological and Anthropocene time-scales, as evaluated through a written test and oral and written examinations
- 7. Devise methods using light stable isotope signatures to trace Earth System interactions across the lithosphere, pedosphere, hydrosphere, biosphere and atmosphere continuum, as evaluated through oral and written examinations.
- Classes are in person: Monday, Wednesday and Friday from 11:30 am 12:20 pm in To be announced

This vast amount of space is reserved for <u>Dy<sup>2</sup>Hr<sup>16</sup>O<sup>32</sup>Sp<sup>1</sup>Here</u>

(First one to solve the riddle and tell Fred by e-mail gets 1 bonus mark)

### Course Outline

This outline is a guide only. *Stable Isotope Science* is a dynamic tool applied to research and scholarship in many different fields. Course content may change depending on class feedback and current topics. If you have interest in a subject not covered here, please contact Fred.

#### 1. Introduction

Importance to earth and environmental sciences, isotopes of interest, some general principles, the atom, chart of the nuclides, atomic mass units, atomic weight, binding energy, nuclear stability, abundance of the elements in the solar system, brief history of stable isotope geochemistry, definitions ( $\delta$ ,  $\alpha$ , 10<sup>3</sup>ln $\alpha$ ,  $\Delta$ ), standards, introduction to analytical methods and instruments (extraction techniques, mass spectrometry), virtual laboratory tour.

### 2. Stable isotopes in the atmosphere and hydrosphere

Equilibrium fractionation of isotopes, kinetic processes, O and H isotopes in water and water vapour, Rayleigh distillation, Global Meteoric Water Line, kinetic isotope fractionation and *d*-excess, evaporation line, regional effects (latitude, altitude, continentality, temperature, precipitation amount), Kuhn, in-cloud processes and O and H isotopes of precipitation, shallow ground water, atmospheric oxygen and carbon dioxide, juvenile water, geothermal water, rock-water ratio, oceanic pore water, formation water, brines (SW Ontario examples), ocean water (salinity, evaporation, dilution, ice-cap effects, ocean currents, climatic effects), snow and ice, marine paleoclimate reconstruction (ice cores, marine foraminifera), terrestrial and lacustrine paleoenvironmental reconstruction (speleothems, mammalian teeth and bones, ostracodes, diatoms), plant phytoliths, tree rings.

### 3. Stable isotopes in igneous rocks

Introduction to igneous rocks, oxygen reservoirs (water, sediments, mantle and derivative rock types), mineral ordering, fractional crystallization, O-isotope geochemistry of granitoid rocks (normal-<sup>18</sup>O, low-<sup>18</sup>O, meteoric water interaction, high-<sup>18</sup>O, role of sediments, isotopic exchange with country rocks), O-isotope geothermometry, high temperature concordancy, retrograde isotopic exchange and disequilibrium, Pegmatite Paradise; meteorites, mass independent fractionation.

### 4. Stable isotopes in sedimentary rocks, weathering and diagenesis

Chemical sediments (carbonate, chert), clastic sediments, submarine weathering (halmyrolysis), Oisotope composition of the ocean through time and its significance, clay mineral structures, controls on clay mineral isotopic compositions during weathering and diagenesis, clay isotope geothermometry, O- and H-isotope variation in soil and weathering clays.

### 5. Stable carbon and nitrogen isotopes in organic matter

Introduction to stable C isotopes on Earth, photosynthesis (C<sub>3</sub>, C<sub>4</sub>, CAM and aquatic plants), Cisotope fractionation during photosynthesis (diffusive, enzymatic), fractionation during organic synthesis, C-isotope behaviour during vegetation shifts and climate change, isotopic alteration of soil organic matter (oxidation, microbial), humic substances, tracking carbon storage in soils (Maya examples), N-isotope variations in air, soil, plants and animals, diet and paleodiet (collagen, structural carbonate in bioapatite, keratin), C- and N- isotope trophic effects, food webs, N-isotope baselines, *are you what you eat*?

### 6. Stable carbon isotopes in the carbon cycle

Carbon cycle (long- versus short-term), carbon reservoirs ( $\delta$  and fluxes), long-term carbon cycle, carbonates, equilibrium C-isotope fractionation, vital effects, coal, petroleum, natural gas, biogenic methane, tracing hydrocarbon leaks, short-term carbon cycle (atmospheric CO<sub>2</sub>, atmosphere-biosphere-hydrosphere transfer, ice-cores, biological pump, ocean productivity), Phanerozoic C-isotope secular variations (atmospheric oxygen, extinction events, Strangelove Ocean, PETM), Earth C-isotope budget.

### 7. Sulphur isotopes

Introduction to S isotopes, fractionation in abiotic and biotic systems, Rayleigh distillation, Recent sediments, ocean water, secular variation, atmospheric sulphur, petroleum, coal, S isotopes as a tracer, mass independent S-isotope fractionation, S isotopes in ore deposits, crystal chemistry and bond-strength controls on S isotopic fractionation, speciation effects on sulphur isotopic compositions ( $f[O_2]$  and pH), comparison with C-isotope system.

### 8. Thinking like an Isotopist

#### Key Sessional Dates:

First Class: Friday, September 5, 2025 Reading Week (no classes): Monday, November 3, 2025, to Sunday, November 9, 2025 Last Class: Monday, December 8, 2025

### Contingency plan for an in-person class pivoting to 100 % online learning

Although the intent is for this course to be delivered in person, should any university-declared emergency require some or all of the course to be delivered online, either synchronously or asynchronously, the course will adapt accordingly. The grading scheme will **not** change. Any assessments affected will be conducted online as determined by the course instructor.

## 4. Course Materials

• <u>Text Book</u>: Sharp, Z. (2<sup>nd</sup> Edition) Principles of Stable Isotope Geochemistry, Electronic Edition.

You can obtain a copy at **no cost** from: <u>csi.unm.edu</u> under the web site's publications tab.

There are assigned readings from this textbook associated with most lectures.

- All course material will be posted to OWL Brightspace: <a href="http://owl.uwo.ca">http://owl.uwo.ca</a>.
- Students are responsible for checking the course OWL Brightspace site (<u>http://owl.uwo.ca</u>) regularly for news and updates. This is the primary method by which information will be disseminated to all students in the class.

 If students need assistance with the course OWL Brightspace site, they can seek support on the <u>OWL</u> <u>Brightspace Help</u> page. Alternatively, they can contact the Western Technology Services Helpdesk. They can be contacted by phone at 519-661-3800 or ext. 83800.

### Technical Requirements (only in case of shift to on-line status)

- Computer and stable internet connection
- If possible, computer with working microphone and/or webcam
- Patience

# 5. Methods of Evaluation

The overall course grade will be calculated as listed below:

Assignments (5)	20 %
Midterm Test	30 %
Interview	15 %
Final Examination	35 %

• Assignments and their due dates will be delivered and received via OWL Brightspace. Normally one week is allowed to complete each assignment. For a few assignments, two weeks are allowed.

- Topics to be included on the Mid-term Test: All material up to the lecture before the Mid-term Test.
- The Mid-term Test will occur during regular class time on **Wednesday, October 29, 2025**. In case of a shift back to 'on-line' learning, the Mid-term Test will still be administered during the regularly scheduled class time. Those details would follow on the course's OWL Brightspace site.
- Student interviews (15 minutes in length) will be scheduled individually and take place between Monday, November 10, 2025, and Friday, November 14, 2025.
- *Topics to be included on the Final Examination:* Entire course but with emphasis on material covered after the Mid-term Test.
- The Final Examination date and location will be scheduled by the Registrar's Office during the normal December examination period. In case of a shift back to 'on-line' learning, the final examination will be administered in a 'take-home' format. Those details would follow on the course's OWL Brightspace site.

### Use of Generative AI Tools – Specific Policy for Earth Sciences 4431a

• In Earth Sciences 4431a, the use of generative AI tools (e.g., ChatGPT, Copilot, Gemini) is **not permitted** in the completion of any course assessments, except for the five assignments, where generative AI is **permitted with limitations**. Those limitations are that the generative AI tool can be used to explore responses to assignment questions but must not be used for final writing of answers. Where a generative AI tool has been used to explore responses to questions, this must be clearly stated with each response where generative AI was used.

#### General information about missed coursework

Students must familiarize themselves with the University Policy on Academic Consideration – Undergraduate Students in First Entry Programs posted on the Academic Calendar:

### https://www.uwo.ca/univsec/pdf/academic policies/appeals/academic consideration Sep24.pdf,

This policy does not apply to requests for Academic Consideration submitted for **attempted or completed work**, whether online or in person.

The policy also does not apply to students experiencing longer-term impacts on their academic responsibilities. These students should consult <u>Accessible Education</u>.

For procedures on how to submit Academic Consideration requests, please see the information posted on the Office of the Registrar's webpage:

#### https://registrar.uwo.ca/academics/academic\_considerations/

All requests for Academic Consideration must be made within 48 hours after the assessment date or submission deadline.

All Academic Consideration requests must include supporting documentation; however, recognizing that formal documentation may not be available in some extenuating circumstances, the policy allows students to make <u>one</u> Academic Consideration request **without supporting documentation** in this course. However, the following assessments are excluded from this, and therefore always require formal supporting documentation:

- Examinations scheduled during official examination periods (Defined by policy)
- Midterm Test (Designated by the instructor as the <u>one</u> assessment that always requires documentation when requesting Academic Consideration)

When a student <u>mistakenly</u> submits their <u>one</u> allowed Academic Consideration request **without supporting documentation** for the assessments listed above or those in the **Coursework with Assessment Flexibility** section below, <u>the request cannot be recalled and reapplied</u>. This privilege is forfeited.

### **Evaluation Scheme for Missed Assessments**

When a student misses the Final Examination and their Academic Consideration has been granted, they will be allowed to write the Special Examination (the name given by the University to a makeup Final Examination). See the Academic Calendar for details (under <u>Special Examinations</u>), especially for those who miss multiple final exams within one examination period.

#### **Essential Learning Requirements**

Even when Academic Considerations are granted for missed coursework, the following are deemed essential to earn a passing grade:

• participation in the interview. When Academic Considerations are granted, the student will be given an opportunity to complete this requirement at a later time

**Assignments.** Students are expected to submit each of the 5 assignments by the deadline listed. Should extenuating circumstances arise, students <u>do not</u> need to request Academic Consideration and they are permitted to submit their assignment up to 24 hours past the deadline without a late penalty. Should students submit their assessment beyond 24 hours past the deadline, a late penalty of 10% per day or fraction of a day will be applied. Assignments received more than 120 hours after the deadline will not be accepted. Academic Consideration requests may be granted only for extenuating circumstances that <u>started before</u> the deadline and <u>lasted longer</u> than the No-Late-Penalty Period (24 hours).

• There are no 'make-up' options for missed Assignments. In case of extenuating circumstances, students should contact the instructor.

# 6. Additional Statements

### 6.1 Religious Accommodation

When conflicts with a religious holiday that requires an absence from the University or prohibits certain activities, students should request an accommodation for their absence in writing to the course instructor and/or the Academic Advising office of their Faculty of Registration. This notice should be made as early as possible but not later than two weeks prior to the writing of the examination (or one week prior to the writing of the test).

Please visit the Diversity Calendars posted on our university's EDID website for the recognized religious holidays:

https://www.edi.uwo.ca.

### 6.2 Accommodation Policies

Students with disabilities are encouraged to contact Accessible Education, which provides recommendations for accommodation based on medical documentation or psychological and cognitive testing. The policy on Academic Accommodation for Students with Disabilities can be found at:

https://www.uwo.ca/univsec/pdf/academic\_policies/appeals/Academic Accommodation\_disabilities.pdf.

### **6.3 Academic Policies**

The website for Registrar Services is <u>https://www.registrar.uwo.ca/</u>.

Use of @uwo.ca email: In accordance with policy,

https://www.uwo.ca/univsec/pdf/policies\_procedures/section1/mapp113.pdf,

the centrally administered e-mail account provided to students will be considered the individual's official university e-mail address. It is the responsibility of the account holder to ensure that e-mail received from the University at their official university address is attended to in a timely manner.

### Requests for Relief (formally known as "appeals")

Policy on Request for Relief from Academic Decision: https://uwo.ca/univsec//pdf/academic\_policies/appeals/requests\_for\_relief\_from\_academic\_decisions.pdf

Procedures on Request for Relief from Academic Decision (Undergraduate): <u>https://uwo.ca/univsec//pdf/academic\_policies/appeals/undergrad\_requests\_for\_relief\_procedure.pdf</u>

### 6.4 Scholastic Offences

Policy on Scholastic Offences: https://uwo.ca/univsec//pdf/academic\_policies/appeals/scholastic\_offences.pdf

Procedures on Scholastic Offences (Undergraduate): https://uwo.ca/univsec//pdf/academic\_policies/appeals/undergrad\_scholastic\_offence\_procedure.pdf

### **Use of Electronic Devices During Assessments**

In courses offered by the Faculty of Science, the possession of unauthorized electronic devices during any in-person assessment (such as tests, midterms, and final examinations) is strictly prohibited. This includes, but is not limited to mobile phones, smart watches, smart glasses, and wireless earbuds or headphones.

Unless explicitly stated otherwise in advance by the instructor, the presence of any such device at your desk, on your person, or within reach during an assessment will be treated as a *scholastic offence*, even if the device is not in use.

Only devices expressly permitted by the instructor (e.g., non-programmable calculators) may be brought into the assessment room. It is your responsibility to review and comply with these expectations.

# • In Earth Sciences 4431a, non-programmable calculators are permitted in written tests and examinations.

### Use of Generative AI Tools – general policy

Unless otherwise stated (see earlier note concerning the five assignments in Earth Sciences 4431a), the use of generative AI tools (e.g., ChatGPT, Microsoft Copilot, Google Gemini, or similar platforms) is **not permitted** in the completion of any course assessments, including but not limited to: assignments, lab reports, presentations, tests, and final examinations.

Using such tools for content generation, code writing, problem solving, translation, or summarization—when not explicitly allowed—will be treated as a **scholastic offence**.

If the use of generative AI is permitted for a particular assessment, the conditions of use will be specified by the instructor in advance. If no such permission is granted, students must assume that use is prohibited. It is your responsibility to seek clarification before using any AI tools in academic work.

### 6.5 Support Services

Please visit the Science & Basic Medical Sciences Academic Advising webpage for information on adding/dropping courses, academic considerations for absences, requests for relief, appeals, exam conflicts, and many other academic-related matters: <u>https://www.uwo.ca/sci/counselling/</u>.

Students who are in emotional/mental distress should refer to Mental Health@Western (<u>https://uwo.ca/health/</u>) for a complete list of options about how to obtain help.

Western is committed to reducing incidents of gender-based and sexual violence (GBSV) and providing compassionate support to anyone who has gone through these traumatic events. If you have experienced GBSV (either recently or in the past), you will find information about support services for survivors, including emergency contacts at:

https://www.uwo.ca/health/student\_support/survivor\_support/get-help.html.

To connect with a case manager or set up an appointment, please contact support@uwo.ca.

Learning-skills counsellors at Learning Development and Success (<u>https://learning.uwo.ca</u>) are ready to help you improve your learning skills. They offer presentations on strategies for improving time management, multiple-choice exam preparation/writing, textbook reading, and more. Individual support is offered throughout the Fall/Winter terms in the drop-in Learning Help Centre, and year-round through individual counselling.

Western University is committed to a thriving campus as we deliver our courses in the mixed model of both virtual and face-to-face formats. We encourage you to check out the Digital Student Experience website to manage your academics and well-being: <u>https://www.uwo.ca/se/digital/</u>.

Additional student-run support services are offered by the USC, <u>https://westernusc.ca/services/</u>.

